

## Chemical Analysis and Biological Activity of Jordanian Chamomile Extracts

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**Abstract:** The Jordanian chamomile (*Matricaria chamomilla*) has been researched more thoroughly to evaluate its useful properties. It is investigated and found that Jordanian chamomile is rich in phenolic compounds, with beneficial biological activities. By applying the most promising HPLC method, the content of total phenolics in methanolic extract was determined according to the Folin-Clocalteu procedure, and was found (GAE>20 mg/g.) The flavonoid types were found as flavones and flavonolos. The minimum inhibitory concentration values for methanolic extracts of Jordanian chamomile were determined for different kinds of bacteria. The extracts have activity against *Staphylococcus aureus*, *Candida albicans*, *Escherichia Coli*, *Betula pubescens* and *Pinus sylvestris*. The activity has been observed to be due to the tannins and a pigenin present in the extract. To utilize these significant sources of natural compounds, further characterization of phenolic composition is needed.

**Key words:** Biological activity, chamomile, extracts, folin-clocalteu procedure, GAE, methanolic, HPLC, phenolic, tannins

### INTRODUCTION

Chamomile is a herb that has been in use since ancient times due to its many advantages and properties (Scalia *et al.*, 1999). Chamomile is said to have antioxidant, anti-inflammatory and anti-bacterial properties. Experts also believe that this wonderful herb has anti-fungal properties too. (Rauha *et al.*, 2002). Chamomile tea is no stranger to us, and we all know the various benefits of chamomile essential oil. Similarly, chamomile extract is also said to be very beneficial, medically. Chamomile extract is in powder form and methanol is used to extract various compounds from the chamomile flowers (Zafari Zangeneh *et al.*, 2010). This extract has been very useful since a long time due to its soothing powers.

Phenolic compounds are commonly found in chamomile, have multiple biological effects (Ramadan *et al.*, 2003). Potential source of antioxidant compounds have been searched in Jordanian chamomile. (Chipault Hawkins, 1952). Flavonoids are especially common in leaves and flowers Chamomile is a well-known and widely used plant for various gastro-intestinal disorders (Burits and Antioxidant, 2000). Extracts of chamomile are used in cosmetics for anti-inflammatory and antiseptic properties (Dick *et al.*, 1987). They may exert antioxidant effects within human body (Maschi *et al.*, 2008). The majority of their antioxidant activity is due to flavones, falvenols, isoflavones, flavonoids, anthocyanin, coumarin, tannins acid, and isocatechins (Svenningsen *et al.*, 2006).

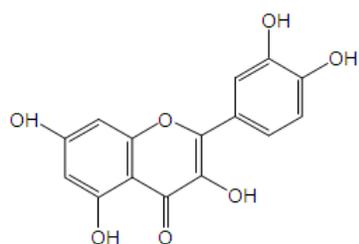
We are interested in plant extraction since phenolics retard oxidative degradation of lipids and improve the

quality of food (Adlercreutz, 1995). phenolic compounds may act as reducing agents, hydrogen donors, and singlet oxygen quenchers. (Rauha *et al.*, 2001). sedative qualities. The scientific explanation for Jordanian chamomile's effectiveness in humans is still being researched; a 2005 study published in the American Chemical Society's "Journal of Agricultural and Food Chemistry" found that volunteers who consumed 5 cups of chamomile tea for two weeks showed an increased level of hippurate. Hippurate is associated with the botanical phenolics that boost immunity by fighting bacteria. This may explain chamomile tea's legendary effectiveness in fighting colds and viruses.

### EXPERIMENTAL

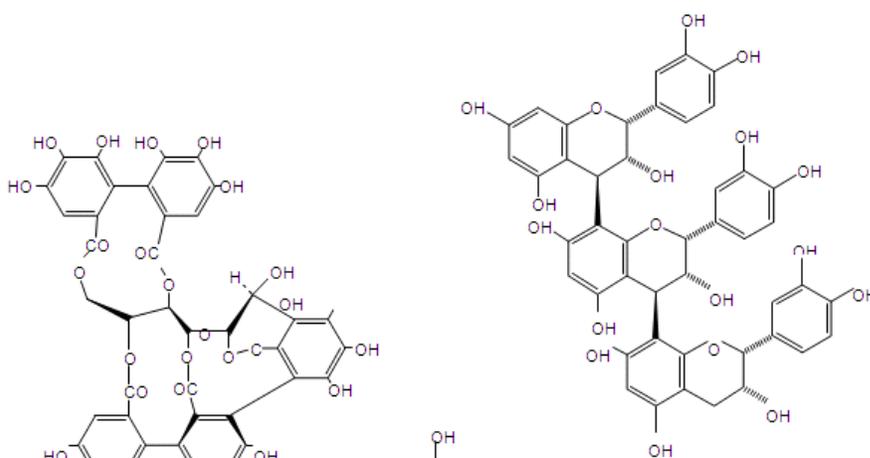
**Preparation of plant extracts:** Dry chamomile flowers of Jordanian origin were purchased from herbal-supermarket at Amman Street summer 2011. Grounded dry chamomile flowers (500 mg) were weighed into a test tube, at laboratories of Basi Science at Applied Science University. A total of 10 mL of 80% aqueous ethanol was added, and the suspension was stirred slightly. Tubes were sonicated 5 min and centrifuged for 10 min. (1500 g), and supernatant were collected. Chamomile was re-extracted twice. Combined supernatant were evaporated to a volume of about 1 mL. These concentrated extracts were lyophilized and weighed.

**Stability and shelf life studied of the extract:** Ethanolic chamomile extracts from the freshly prepared were transferred in 1.5 mL eppendorf tubes.



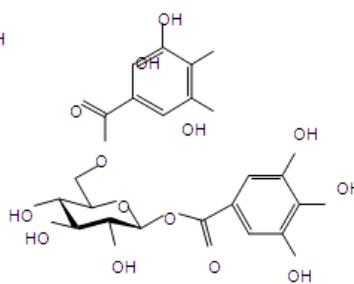
Quercetin  
flavonol

a) flavonoids



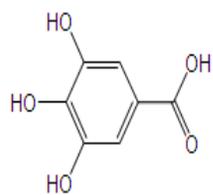
Castalagin  
ellagitannin

Procyanidin C-1  
condensed tannin

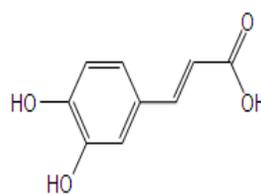


E-1,6-digalloyl-D-  
glucose gallotannin

b) tannins



Gallic acid  
3,4,5-trihydroxybenzoic acid



Caffeic acid  
3,4-dihydroxycinnamic

c) Phenolic acids

Fig. 1: Typical structures of some groups of plant phenolic

These tubes were distributed in 3 groups to store at -20°C, 4°C and at room temperature. The tubes were retracted at 15, 30, 60, 90 and 120 days for analysis and dissociation

of glucoside bond and. In addition, we performed stability studies on ethanolic chamomile extracts at different pH, light exposure and long term storage.

Table 1: Composition of Jordanian chamomile extract (mg/g of dried extract)

| Compounds     | Concentration mg/g | Retentions time (min) |
|---------------|--------------------|-----------------------|
| Quercitrin    | 23.7               | 6.5                   |
| Myricitrin    | 7.9                | 16.5                  |
| Kaempferitrin | 10.1               | 18.5                  |

Table 2: Minimum Inhibitory Concentration (MIC) of chamomile extract against different microorganism

| Microorganism           | MIC of extract $\mu\text{g/mL}$ |
|-------------------------|---------------------------------|
| <i>Staphylococcus</i>   | 32.5                            |
| <i>Candida albicans</i> | 65                              |
| <i>Escherichia Coli</i> | 250                             |
| <i>Betula pubescens</i> | 125                             |
| <i>Pinus sylvestris</i> | 500                             |

**Determination of total phenolics:** The amount of total phenolics in extracts was determined according to Folin-Ciocalteu procedure (10). The total phenolic content was expressed as Gallic Acid Equivalents (GAE) in mg/g dry material.

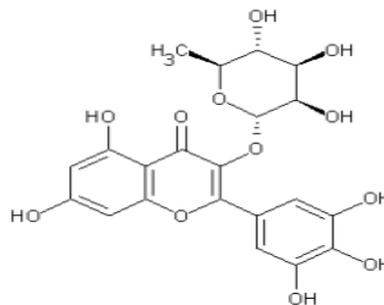
**HPLC analysis:** A 10  $\mu\text{L}$  of methanolic solution was injected into HPLC system (GRD-10-0062) with a UV-visible detector. A Cap cell Pack UG 120C 18 columns was used. The mobile phase consisted of 80% methanol. The gradient was 20% in 60 min at a flow rate 0.20 mL/min.

**Minimum Inhibitory Concentration (MIC) test:** The samples were tested for their antimicrobial testing in vitro by the agar dilution technique. All samples were dissolved in Dimethyl Sulphoxide Solvent (DMSO) for the antimicrobial test and the solutions were sterilized by membrane filtration. Aliquots of samples were diluted with melted tryptic Soy agar, tryptone, soytone, sodium chloride and agar to give concentrations of 2000, 1500, 1000, 500, 250, 125, 62.5 and 31.3  $\mu\text{g/mL}$ . The MIC value were noted after 24 h at 37°C as shown in Table 2.

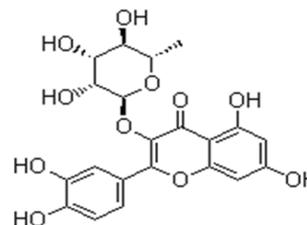
## RESULTS AND DISCUSSION

The amount of dry weight of extract was 17mg, while the amount of total phenolics in chamomile extract was 23.2 mg/g. UV-vis Absorption at 280-530 nm were measured (PerkinElmer spectrophotometer Norwalk, CT), p-hydroxybenzoic acid can be detected at 260nm, catechins at 280nm, and anthocyanins at 530nm (Fig. 3). The conjugated C-C double bonds which act as chromophores made (UV/Vis) detector can be used in both qualitative and quantitative analysis of nature-derived samples containing phenolics (Fig. 1 and Table 1).

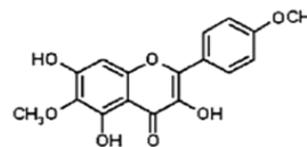
Identified compounds in extract were: Kaempferitrin (at 320 nm), Quercitrin, Myricitrin (at 280 nm). Figure 2, but no identified substances as Anthocyanins. The phenolic compounds of natural origin have the positive



Myricitrin



Quercitrin



Kaempferitrin

Fig. 2: Structure of isolated compounds

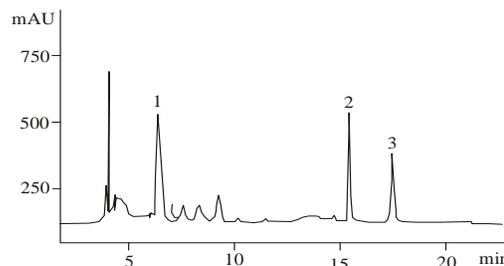


Fig. 3: HPLC-UV profile of methanol extract, detection at 280 nm. 1: Kaempferitrin; 2: Quercitrin; 3: Myricitrin

property of being soluble in polar solvents, this leads to the possibility of using reversed phase HPLC, stationary phase was octadecylsilane (ODS, C18, RP-18), The eluent was methanol and (UV/Vis) detector (Fig. 3).

These glucosides are highly stable in solution at different temperature range and their degradation occurs after long-term storage and extraction conditions at different pH and solvent.

The extract was tested for its antimicrobial testing *in vitro* by the agar dilution technique (Adlercreutz, 1995) the extract of chamomile has activity against *Staphylococcus* and *Candida albicans*, and noticeable activity against other microorganisms (Table 2).

## CONCLUSION

Three different compounds were isolated and identified by HPLC technique as flavone and flavonol, Compounds are; kaempferol-o-glucosides, quercetin-o-glycosides, and myricetin-o-glycosides, but no identified compound as anthocyanins. The chamomile extract showed antimicrobial activity in this study, Jordan provides an interesting source of plants for studying their chemical composition and its effects in biological systems and man. Several serviceable *in vitro* tests that illuminate the property of interest are available for screening plants and their constituents in order to find the most effective materials and components for further investigations.

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